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(54) MATERIAL FOR ABSORBING AQUEOUS FLUIDS.

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Description

The present invention relates to water-swella-
ble polymers which are useful in absorbing aqueous
fluids.

Water absorbent materials comprising water-
swella-ble polymers have provided various uses in
the art. See, for example, U.S. Patent Nos.
3,926,891; 4,190,562; 4,293,609 and 4,424,247. Nu-
merous disposable articles containing such water-
swella-ble polymers are disclosed as in U.S. Patent
Nos. 3,669,103 and 3,888,257.

Water-swella-ble polymers which are useful in
providing absorbent materials are typically poly-
mers comprising crosslinked acrylamide; hydrolyz-
ed acrylamide; acrylic acid; hydrolyzed acrylates;
hydrolyzed acrylonitrile; and grafted starch and cel-
luloses; and the like. Such polymers require con-
siderable care in controlling the amount of crosslin-
king which is present. For example, if the amount
of crosslinking in the polymer is increased, the gel
strength of the resulting polymer network is in-
creased at the expense of reducing the capacity of
the gel to absorb fluid. On the other hand, at low
degrees of crosslinking the gels have a poor, slimy
consistency which exhibit low gel strengths.

In U.S.-A-4,432,881, for example, there is dis-
closed a water-dispersible thickening agent which
dissolves into water or an aqueous medium when
brought into contact therewith, thus forming a vis-
cous solution having a very low gel strength.

U.S.-A-4,304,902 relates to crystalline polymers
on the basis of ethylene oxide with a minor amount
of a second alkylene epoxide which have good
solubility in water, and form viscous solutions that
retain their viscosity under shear stress and at
elevated temperatures.

U.S.-A-4,062,817 and 4,190,562 teach poly-
merisation in a surfactant-free medium to obtain
water absorbent polymers, whereas according to
US-A-4,066,583 water absorbent polymer composi-
tions in form of flexible films, fibers, and the like
can be obtained by copolymerizing potassium, am-
monium and amine salts of unsaturated carboxylic
acids with alkylacrylates or methacrylates having
10 to 30 carbon atoms and at least one alkyl-
acrylate or methacrylate with 1 to 9 carbon
atoms in the alkyl group, an acrylic or methacrylic
nitrile or amide and mixing the polymer so ob-
tained with a considerable amount (>30 weight
parts per 100 of polymer) of an aliphatic diol.

In view of the fact that the utility of known
water-absorbent materials is limited due to lack of
integrity of the gel structure during use, insufficient
absorbing capacity and absorbing rate, and difficul-
ties in crosslinking the polymers which are em-
ployed; it would be highly desirable to provide a
novel, highly absorbent polymer system having a

high gel strength and minimal amounts of crosslin-
king.

The present invention is a composition capable
of absorbing an aqueous liquid which composition
comprises a functionally effective amount of a
swella-ble addition polymer comprising in polymer-
ized form.

(a) a major portion of at least one water-solu-
ble, ionizable or hydrolyzable monomer which is,
capable of forming at least a 10 weight percent
solution in water and which readily undergoes
addition polymerisation to form water-soluble
polymers.

(b) a minor portion of at least one water-insolu-
ble ethylenically unsaturated monomer having a
pendant hydrophobic moiety where less than 0.4
weight parts of the water-insoluble monomer will
dissolve in 100 parts by weight water. and

(c) a minor portion of a crosslinking moiety in an
amount sufficient to provide gel strength to the
polymer with the proviso that the composition
does not comprise an interpolymer of an al-
iphatic glycol having 2 - 10 carbon atoms, and
when the polymerized monomers comprise an
acrylic ester with an alkyl group having 10 to 30
carbon atoms, the polymer does not comprise
polymerized monomers of an acrylic acid ester
with an alkyl group having 1 - 9 carbon atoms,
or an acrylic or methacrylic nitrile or amide and
which polymer is capable of undergoing hy-
drophobic association through hydrophobic
groups of said polymer such that said polymer
is swella-ble but not readily soluble in the pres-
ence of an aqueous liquid when said polymer is
in intimate contact with an amount of surfactant,
if any, which is less than that amount of surfac-
tant which will solubilize said polymer in said
aqueous liquid.

The composition of the present invention may
additionally contain a functionally effective amount
of an inorganic colloidal support.

The compositions of this invention, which have
absorbed aqueous fluids, exhibit gel strengths
which can increase with time. That is, swollen gel
particles of this invention in contact with one an-
other are capable of forming one large continuous
gel particle by what is believed to be intergel
associations. This expected change in structure of
the gel provides an increase in gel strength of the
gel as a whole. In addition, the formation of a large
gel particle provides a gel having less tendency to
migrate during use. Gels of this invention have
improved adhesive and elasticity characteristics.

The compositions of this invention are capable
of absorbing many times their own weight of an
aqueous fluid. That is, the compositions of this
invention can have improved aqueous fluid absorb-
ing capacity, exhibit improved rates of aqueous

fluid absorption, and have improved gel strengths. Consequently, the compositions are useful in providing a process for absorbing large amounts of an aqueous fluid. The compositions are useful in a wide variety of applications wherein water-absorbent polymeric materials have been used.

By "hydrophobic associative" is meant that, in aqueous media, the pendant hydrophobic moieties of the polymer associate thereby preventing solubilization and providing swellability of the polymer in an aqueous medium.

The hydrophilic/hydrophobic polymer (i.e., water swellable addition polymer) is predominantly hydrophilic, but contains a sufficient concentration of suitable hydrophobic moieties to enable it to associate with other hydrophobic moieties and the hydrophobic moieties of any surfactant which is present to cause a substantial swellability of the polymer in the presence of an aqueous medium. That is, the hydrophilic monomers polymerize to form a major portion of the polymer and the hydrophobic monomers polymerize to form a minor portion of the polymer. The polymer has an unsassociated molecular weight which can vary such that the polymer will form a gel. Polymers having very high molecular weight, e.g., weight average molecular weight greater than 5 million, are believed to be suitably employed. Preferably, the hydrophilic/hydrophobic polymers used in this invention have weight average molecular weights in the range from 200,000 to 10 million, most preferably from 800,000 to 5 million. See U.S. Patent No. 4,432,881.

The hydrophobic groups of the hydrophilic/hydrophobic polymer are preferably pendant organic groups having hydrophobicities comparable to one of the following: aliphatic hydrocarbon groups having at least four carbon atoms such as C_4 to C_{20} alkyls and cycloalkyls; aromatic and aromatic hydrocarbon groups such as naphthyls; alkylaryls wherein alkyl has one or more carbon atoms, preferably 4 to 8 carbon atoms; haloalkyls of 4 or more carbon atoms, preferably perfluoroalkyls;

Suitable hydrophobic monomers include those which are (1) water-insoluble, i.e., where less than 0.4, preferably less than 0.2, weight part of the hydrophobic monomer will dissolve in 100 weight parts water and (2) ethylenically unsaturated compounds having hydrophobic groups as defined hereinbefore. Exemplary hydrophobic monomers include the higher alkyl esters of α,β -ethylenically unsaturated carboxylic acids such as dodecyl acrylate, dodecyl methacrylate, tridecyl acrylate, tridecyl methacrylate, tetradecyl acrylate, tetradecyl methacrylate, octadecyl acrylate, octadecyl methacrylate, ethyl half ester of maleic anhydride, diethyl maleate, and other alkyl esters

derived from the reactions of alkanols having from 4 to 20, preferably from 8 to 20, carbon atoms with ethylenically unsaturated carboxylic acids such as acrylic acid, methacrylic acid, fumaric acid, itaconic acid and aconitic acid, and maleic anhydride; alkylaryl esters of ethylenically unsaturated carboxylic acids such as nonyl- α -phenyl acrylate, nonyl- α -phenyl methacrylate, dodecyl- α -phenyl acrylate and dodecyl- α -phenyl methacrylate; N-alkyl ethylenically unsaturated amides such as N-octadecyl acrylamide, N-octadecyl methacrylamide, N,N-diethyl acrylamide and similar derivatives thereof; α -olefins such as octene-1, decene-1, dodecene-1 and hexadecene-1; vinyl alkylates wherein alkyl has at least 8 carbons such as vinyl laurate and vinyl stearate; vinyl alkyl ethers such as dodecyl vinyl ether and hexadecyl vinyl ether; N-vinyl amides such as N-vinyl lauramide and N-vinyl stearamide; and ar-alkylstyrenes such as t-butyl styrene. of the foregoing hydrophobic monomers, the alkyl ester of acrylic acid, methacrylic acid, N-alkyl acrylamides and N-alkyl methacrylamides wherein alkyl has from 8 to 20 carbon atoms, styrene and the alkyl styrenes wherein alkyl has from 4 to 8 carbons such as t-butyl, are preferred. The alkylmethacrylates and alkylacrylates wherein alkyl has from 10 to 20 carbon atoms are more preferred. Dodecyl methacrylate and N-dodecyl methacrylamide are the most preferred where hydrolysis is not a problem.

The water-soluble monomers are those ionizable or hydrolyzable monomers which are sufficiently water-soluble to form at least a 10 weight percent solution when dissolved in water and readily undergo addition polymerization to form polymers which are water-soluble. Exemplary water-soluble monomers include ethylenically unsaturated carboxylic acids such as acrylic acid, methacrylic acid, itaconic acid and fumaric acid; ethylenically unsaturated amides such as acrylamide, methacrylamide and fumaramide and their N-substituted derivatives such as 2-acrylamido-2-methylpropane sulfonic acid (AMPS), N-(dimethylaminomethyl)acrylamide as well as N-(trimethylammoniummethyl)acrylamide chloride and N-(trimethylammoniumpropyl)methacrylamide chloride; and other ethylenically unsaturated quaternary ammonium compounds such as vinylbenzyl trimethyl ammonium chloride; sulfoalkyl esters of unsaturated carboxylic acids such as 2-sulfoethyl methacrylate, aminoalkyl esters of unsaturated carboxylic acids such as 2-aminoethyl methacrylate; vinyl amines such as vinyl pyridine and vinyl morpholine; diallyl amines and diallyl ammonium compounds such as diallyl dimethyl ammonium chloride, vinyl heterocyclic amides such as vinyl pyrrolidone, vinylaryl sulfonates such as vinylbenzyl sulfonate; as well as the salts of the foregoing

monomers. Preferable salts include the sodium, potassium and ammonium salts of the monomers. Of the foregoing water-soluble monomers, acrylic acid, including salts of acrylic acid such as sodium acrylate or ammonium acrylate, and acrylamide and combinations of these are preferred. Hydrolyzable monomers can also be employed.

The hydrophilic/hydrophobic polymer is preferably an addition copolymer of a water-soluble ethylenically unsaturated monomer and an ethylenically unsaturated monomer having sufficient concentration of hydrophobic groups to enable the resulting polymer to form a gel under conditions of use. Exemplary preferred polymers include copolymers of from 90 to 99.995, more preferably from 98 to 99.995, most preferably from 99 to 99.9, mole percent of one or more water-soluble monomers with from 0.005 to 10, more preferably from 0.005 to 2, most preferably from 0.1 to 1, mole percent of one or more hydrophobic monomers. For these polymers, preferred amounts of hydrophobic monomers can vary with the molecular weight of the polymer.

The crosslinking of the monomers can be provided by a variety of means. See, for example, U.S. Patent No. 4,293,609. The amount of crosslinking agent can vary and can suitably range from 0 to 5 weight percent, preferably from 300 to 3,000 ppm of crosslinkable monomer, based on all polymerized monomers. Preferably, the crosslinking agent is a polyvinyl monomer. Examples of polyvinyl monomers include divinyl benzene, acryloyl or methacryloyl polyesters of polyhydroxylated compounds, divinyl esters of polycarboxylic acid, diallyl esters of polycarboxylic acids, diallyl dimethyl ammonium chloride, triallyl terephthalate, methylene bisacrylamide, diallyl maleate, diallyl fumarate, hexamethylene bis maleimide, triallyl phosphate, trivinyl trimellitate, divinyl adipate, glyceryl trimethacrylate, diallyl succinate, divinyl ether, the divinyl ethers of ethylene glycol or diethylene glycol diacrylate, polyethylene glycol diacrylates or methacrylates, 1,6-hexanediol diacrylate, pentaerythritol triacrylate or tetracrylate, neopentyl glycol diacrylate, cyclopentadiene diacrylate, the butylene glycol diacrylates or dimethacrylates, and trimethylolpropane di- or tri-acrylates.

The polymers of this invention can be prepared using aqueous solution techniques, inverse bead suspension techniques, spray bead polymerization techniques using UV catalysis, or other such techniques. Preferred methods of preparation include the aqueous solution techniques.

The aforementioned hydrophilic/hydrophobic polymers containing hydrophobic moieties are advantageously prepared by copolymerizing the water-soluble monomers with hydrophobic monomers in an aqueous medium containing an ionic

surfactant or emulsifier that solubilizes the hydrophobic monomer in the aqueous medium. By solubilizing the hydrophobic monomer in the aqueous medium, the combination of aqueous medium, hydrophobic monomer and surfactant gives the visual appearance of a clear or translucent solution. That is, a micelle forming amount of a surfactant is employed. This copolymerization is preferably accomplished in an aqueous medium containing a polymerization initiator capable of generating free-radicals. Optionally, a chain transfer agent may be included in the polymerization reaction mixture.

The solubilizing surfactant is required in most instances to suitably solubilize the hydrophobic monomer and to subsequently obtain a hydrophilic/hydrophobic polymer having a desirable concentration of hydrophobic moieties in the polymer. It is believed that the hydrophobic monomer is solubilized in the micelles formed by the surfactant. Thus, the surfactant is generally employed in an amount which is above the critical micelle concentration (CMC) of the surfactant or surfactant mixture, but less than that which reduces the concentration of hydrophobic monomer in the resultant hydrophilic/hydrophobic polymer to a point that the polymer will not form a gel under the conditions in which the product is used. The amount of surfactant used will also be such that there is believed to be at least one hydrophobic monomer molecule per micelle of the surfactant. An upper limit to the amount of hydrophobic monomer which is employed is based on that level which can be solubilized by the surfactant, i.e., an amount of hydrophobic monomer below which the reaction mixture becomes cloudy. For example, when sodium dodecyl sulfate (NaDS) is employed as a surfactant for dodecyl methacrylate (DMA), the molar ratio of DMA to NaDS is from at least 1:50 up to 1:2, preferably from 1:5 to 1:25, most preferably from 1:10 to 1:15. By knowing the CMC, the hydrophilic-lipophilic balances (HLB) and aggregation number of molecules in the micelle molecular weight of a surfactant and the hydrophobicity of the hydrophobic monomer, suitable molar ratios and appropriate surfactant concentrations can be determined for any given hydrophobic monomer and surfactant to provide similar suitable concentrations of hydrophobic moieties in the hydrophilic/hydrophobic polymer.

Suitable surfactants include, for example, anionic agents such as alkali metal salts of alkyl sulfates and alkyl and aryl sulfonates, e.g., dodecyl alkyl sulfosuccinates and sodium dodecylbenzene sulfonate; fatty acid soaps, e.g., sodium oleate, sodium stearate and potassium oleate; alkali metal salts of sulfonated fatty alcohols, e.g., sodium dodecyl sulfate; sulfates of ethoxylated alcohols; alkyl phosphate esters, e.g., dodecyl hydrogen

phosphate; and fluoro emulsifiers, e.g., perfluoroalkyl sulfates. Also included are cationic surfactants such as alkylamine hydrochlorides, e.g., dodecylamine hydrochloride and tridecylamine hydrochloride; quaternary alkyl or aryl ammonium halides such as dodecyl trimethyl ammonium chloride; ethoxylated fatty amines and other surfactant as described in McCutcheon's Detergents and Emulsifiers, North American Edition, 1980 Annual. In general, when the hydrophilic/hydrophobic polymer is anionic or nonionic, an anionic surfactant such as an alkali metal alkyl sulfate is preferably employed as the surfactant. When the hydrophilic/hydrophobic polymer is cationic, a cationic surfactant such as dodecylamine hydrochloride is preferably employed. When the hydrophilic/hydrophobic polymer is nonionic, anionic or cationic, a nonionic surfactant such as nonylphenoxy polyethylene glycol having 10 ethyleneoxy units per molecule or other water-dispersible nonionic surfactants as defined herein can be employed subject to the condition that an ionic surfactant be employed.

Exemplary suitable polymerization initiators include the azo catalysts such as azobisisobutyronitrile, 2,2'-azobis(2,4-dimethylpentanenitrile) and dimethyl azoisobutyrate; organic peroxygen compounds such as benzoyl peroxide, t-butyl peroxide, diisopropyl benzene hydroperoxide and t-butyl hydroperoxide; and inorganic persulfates such as potassium persulfate, ammonium persulfate and sodium persulfate. Of these initiators, the oil-soluble types such as the organic peroxides and azo compounds are preferred. It is desirable to employ from 0.01 to 0.1 weight percent of initiator based on the monomers.

In an optional embodiment, a nonionic surfactant is added to the polymerization recipe as another component in addition to the ionic surfactant, i.e., emulsifier. Alternatively, it is suitable to add the nonionic surfactant to the hydrophilic/hydrophobic polymer subsequent to polymerization. Exemplary nonionic surfactants include the reaction products of ethylene oxide or mixtures of ethylene oxide and higher alkylene oxide with active hydrogen compounds such as phenols, alcohols, carboxylic acids and amines, e.g., alkylphenoxyethyleneoxy ethanols. More preferred nonionic surfactants are the alkyl polyethyleneoxy compounds represented by the formula:



wherein R is C₈-C₁₈ alkyl, EO is ethyleneoxy and n is a number from 1 to 10. Other suitable nonionic surfactants are described in McCutcheon's, *supra*. Of the foregoing surfactants, the ethoxylated alkyl phenol and ethoxylated fatty alcohols are more

preferred.

Suitable anionic substituted polyethyleneoxy compounds useful herein are represented by the formula:



wherein R and n are as defined hereinbefore. EO is ethyleneoxy and X is SO₃H or CH₂CO₂H or PO₃H; salts of long chain carboxylates such as, for example, potassium oleate, sodium laurate, potassium stearate, potassium caprolate, and sodium palmitate; alkali metal alkylbenzene sulfonates such as sodium nonylbenzene sulfonate and potassium dodecylbenzene sulfonate; alkali metal alkyl sulfates such as sodium dodecyl sulfate and alkali metal dialkyl sulfosuccinates such as sodium dihexyl sulfosuccinate and sodium dioctyl sulfosuccinate; salts of resin acids such as abietic acid and dihydroabietic acid.

The hydrophilic hydrophobic polymers when made using unneutralized acidic monomers are conveniently neutralized using organic or inorganic bases, by mixing the polymer with the base. Bases such as, for example, ammonia, ammonium hydroxide, alkali metal hydroxides, alkali metal bicarbonates, carbonates, alkali metal salts of basic phosphates, silicates, organic amines such as alkyl amines, and alkanol amines can be used. It is also understood that the polymer can be neutralized after recovering the polymer, if so required. The degree of neutralization in the polymers can vary from 0 to 100 percent, preferably from 50 to 80 percent.

The hydrophilic hydrophobic polymers are readily recovered from the aqueous medium when such is desired by removal of water under vacuum or by azeotropic distillation or by drum drying. The polymer can be ground to the desired size (e.g., such as a powder, prior to use).

It is also understood that hydrophilic/hydrophobic polymers of monomers such as acrylamide or methylacrylate, acrylic acid and hydrophobic monomer can be prepared by copolymerizing all three of these monomers or by copolymerizing acrylamide or methylacrylate with the hydrophobic monomer and subsequently hydrolyzing a portion of the copolymerized hydrolyzable monomer such as acrylamide or methylacrylate by contacting the copolymer with a base such as sodium hydroxide and/or sodium carbonate.

The inorganic colloidal support is preferably a water-insoluble material which has a particle size small enough that the particles disperse in an aqueous solution to yield transparent or translucent fluids. Typical colloidal particles which are useful as colloidal supports are inorganic materials having

a particle diameter from 5 to 20 nm (50 Å to 200 Å), preferably from 8 to 10 nm (80 Å to 100 Å). Clusters of particles are not generally larger than 100 nm (1,000 Å) in diameter. Colloidal supports can be made of any substantially water-insoluble inorganic materials. Suitable materials include, for example, colloidal or fumed silica, alumina, and titania as well as colloidal oxide solutions such as those containing, for example, zinc, zirconium, nickel, iron, or cobalt.

The polymer can be blended with the colloidal support using a variety of techniques. For example, the support and polymer can be dry blended. Most preferably, the support is dispersed in an aqueous liquid and contacted with the polymer, which is either in a gel or a dry state. The support and polymer mixture can be dried, if desired. It is also possible to polymerize the monomers in the presence of a dispersed colloidal support.

The amount of colloidal support which is employed can vary. Suitable amounts of the colloidal support range from 0 to 80, preferably from 0.2 to 50, weight percent based on the weight of polymer and support. Such compositions can exhibit increased gel strengths over those compositions not containing the colloidal support.

The compositions of this invention can be further, though optionally, blended with a desired inorganic filler which can be non-colloidal in nature. Although the materials can be physically blended using a wide variety of means, dry blending is preferred. That is, an essentially dry polymer is physically mixed with the desired filler which is an essentially dry inorganic powder. Thorough mixing insures good gel strength of the composition. Generally the order of addition of components is not particularly critical. More preferably, the filler is added to the components present in the polymerization recipe before total polymerization has occurred.

By the term "filler" is meant a very finely divided water insoluble or sparingly water-soluble grouping or aggregate of solid particles, usually smaller than 1,000 μm but which is not generally colloidal in nature. Inorganic powders especially preferred herein include alumina trihydrate and a wide variety of clays. Specific examples include sodium bentonite (montmorillonite clay), kaolinite and attapulgite. Other inorganic powders include, for example, white carbon, synthetic silicate white carbon, basic magnesium carbonate, ultrafine magnesium silicate, light or heavy calcium carbonate, soft or hard clays, talc, vermiculite, pearlite, barium sulfate, and mica.

The amount of polymer and optional colloidal support employed in blends with filler can preferably range from 20 to 100, more preferably from 40 to 75, and most preferably from 60 to 70, weight

percent of polymer and colloidal support based on the total weight of the polymer, colloidal support and filler. Conversely, the amount of filler useful herein preferably ranges from 0 to 80, more preferably from 25 to 60, and most preferably from 30 to 40, weight percent of said filler based on the total weight of the polymer, colloidal support and filler.

Such blended compositions (i.e., containing polymer, optional colloidal support and filler) can exhibit further increased gel strengths over those compositions containing a polymer and not containing the inorganic filler. The high gel strengths of such compositions allow for products exhibiting good utilization of water absorbent capacity as well as good integrity. In this regard, the compositions of this invention are easy to handle and can be employed in a wider variety of applications compared to previously employed water absorbent materials.

The compositions of this invention can be incorporated into film laminates and other such materials as are described, for example, in U.S. Patent Nos. 4,117,184; 4,176,677; 4,293,609 and 4,424,247. For example, polymer compositions can be incorporated in wicking substrates and treated as described in U.S. Patent No. 4,293,609. Such materials exhibit high water absorbent capabilities. The compositions of this invention and the laminates prepared therefrom can be incorporated into absorbent devices. The compositions can be used in those applications which have employed hydrophilic polymers as highly absorbent materials. The compositions can be employed in conjunction with water-absorbent polymeric materials known in the art, as for example, polymeric blends.

The compositions of this invention can be shaped and formed for use in a wide variety of applications. For example, the compositions can be formed into sheets, filaments, coatings or molded articles. The compositions can be foamed, extruded into articles or extruded into a pellet form. The compositions of this invention can be incorporated into a wide variety of product forms.

Uses of the increased gel strength composition of the present invention include applications such as, for example, shut-off filters for fuels such as gasoline, oil, and hydraulic fluids. Other applications include the incorporation of such compositions, as is necessary, in disposable diapers, tampons, and other personal hygiene products and applications.

The following examples are presented to further illustrate but not limit the scope of this invention. All parts and percentages are given based on weight, unless otherwise noted.

Example 1

A clear mixture of 1.05 grams (g) lauryl methacrylate in 12.5 g of a 30 percent solution of sodium lauryl sulfate is charged to a one-half liter jacketed reactor equipped with a stirrer, nitrogen sparge tube, gas outlet and thermometer. To the reactor is charged 270 g deionized water, 36 g acrylic acid, and 0.024 g of active 2,2'-azobis(2,4-dimethylpentanenitrile) dissolved in 1 g polyethylene glycol of secondary alcohol surfactant. The reaction mixture is sparged with nitrogen. The reactor jacket is brought to 60°C and this temperature is maintained for 4 hours. The resulting gel is neutralized using 22.8 g of a 28.4 percent active aqueous ammonia solution. The neutralized gel is dried on a steam heated rotating drum apparatus. The resulting flake material is ground in a Waring Blendor, Trademark of Dynamics Corporation of America, and screened such that particles having a size of less than 0.84 mm (20 mesh) are obtained.

The free swell capacity of the polymer (i.e., the grams of fluid absorbed per gram of polymer) is determined by dispersing 0.5 g of polymer in 150 g of a 1 percent aqueous sodium chloride solution, waiting 20 minutes, removing nonabsorbed water by pouring the mixture into a funnel lined with 150 nm (150 micron) nylon screen, allowing free water to drain for 20 minutes, and weighing the drained (i.e., nonabsorbed) water. After drainage the gel is tough and firm indicating good gel strength. The free swell capacity of the product is 54 gram of water absorbed per gram of polymer.

Example 2

A mixture of 1.05 g lauryl methacrylate in 6.25 g of a 60 percent solution of sodium dodecyl benzene sulfonate is charged into a reactor, as described in Example 1. To the reactor is charged 306 g deionized water, 36 g acrylic acid and 0.048 g of active catalyst dissolved in 1 g surfactant, as described in Example 1. The mixture is subjected to polymerization conditions and isolated, as in Example 1. The free swell capacity of the product is 97 gram of water absorbed per gram of polymer, as determined as described in Example 1.

Example 3

A mixture of 0.75 g stearyl methacrylate in 6 g of a 30 percent solution of sodium lauryl sulfate is charged into a reactor, as described in Example 1. To the reactor is charged 315 g deionized water, 36 g acrylic acid and 0.024 g of active catalyst dissolved in 1 g surfactant, as described in Example 1. The mixture is subjected to polymerization conditions and isolated, as in Example 1. The free swell capacity of the product is 91 gram of water absorbed per gram of polymer, as determined as

described in Example 1.

Example 4

A mixture of 0.63 g lauryl methacrylate in 6.25 g of a 30 percent solution of sodium lauryl sulfate is charged into a reactor, as described in Example 1. To the reactor is charged 275 g deionized water, 36 g acrylic acid and 0.038 g of active catalyst dissolved in 1 g surfactant, as described in Example 1. To the mixture is added 15.4 g of alumina trihydrate. The mixture is subjected to polymerization conditions and isolated, as in Example 1. The free swell capacity of the product is 70 gram of water absorbed per gram of polymer, as determined as described in Example 1.

Example 5

Into a citrate bottle is charged 96.6 parts acrylic acid (99.0 mole percent of monomers) as a 40 percent aqueous solution. The citrate bottle contains 10 parts sodium lauryl sulfate added as a 10 percent aqueous solution, 9.6 parts of a nonionic surfactant represented as $C_{12}H_{25}(EO)_5-H$, 6.4 parts of a nonionic surfactant represented as $C_{12}H_{25}(EO)_{10}-H$, and 3.4 parts (1.0 mole percent of monomers) lauryl methacrylate. The solution is stirred and 500 parts per million (ppm) based on total weight of the monomers of methylene-bis-acrylamide is added as an aqueous solution having a concentration of 1.25 percent in water. To this mixture is added 1,000 ppm based on monomer of an azobisisobutyronitrile catalyst as a 1.25 percent solution in t-butyl alcohol. Distilled water is added to bring the monomer concentration of the mixture to 10 percent. The bottle is purged with nitrogen and the bottle is capped. The contents are subjected to polymerization conditions at 60°C for 18 hours. The product is a thick gel which is drum dried and ground to 0.84 mm (20 mesh) particle size. The product is designated as Sample A. Gel capacity of this product is determined to be 70 g of absorbed 1 percent aqueous sodium chloride solution per gram of product. Gel capacity is determined by contacting about 0.2 g of product with 25 g of said solution for about 20 minutes. The mixture is drained on 0.84 mm (20 mesh) screen and weighed.

Gel strength of this wet gel is determined to be 7.6 kPa (1.1 pounds per square inch). Gel strength is determined by placing weight on a piston placed over the gel until gel begins to extrude through the 0.84 mm (20 mesh) screen that the gel rests on.

For comparison purposes, an acrylic acid polymer crosslinked with 500 ppm methylene-bis-acrylamide yields a slimy product upon contact with the sodium chloride solutions and exhibits a gel

strength of less than 2.8 kPa (0.4 pounds per square inch).

Example 6

A product is prepared and evaluated as described in Example 5, except that the recipe includes 1,000 ppm methylene-bis-acrylamide rather than 500 ppm. The product is designated Sample B. The gel capacity of the product is 45 g of absorbed 1 percent aqueous sodium chloride solution per gram of product. The gel strength of the product is 7.6 kPa (1.1 pounds per square inch), determined as described in Example 5.

For comparison purposes, an acrylic acid polymer crosslinked with 1,000 ppm methylene-bis-acrylamide yields a slimy product upon contact with the 1 percent sodium chloride solution, and exhibits a gel strength of less than 2.8 kPa (0.4 pounds per square inch).

Example 7

To each of 0.2 g of product Samples A and B from Examples 5 and 6 are added 2.5 g of a 2 percent dispersion of colloidal silica in a 1 percent aqueous sodium chloride solution. The product and dispersion are mixed, and to the mixture is added a few drops of ammonium hydroxide solution. Gel capacity of the sample is determined as described in Example 5, and is determined to be 75 g of absorbed 1 percent aqueous sodium chloride solution per gram of polymer for Sample A mixed with the colloidal silica. The gel strength of Sample A mixed with the colloidal silica is 19.0 kPa (2.75 pounds per square inch). Gel capacity is determined to be 80 g of absorbed 1 percent aqueous sodium chloride solution per gram of polymer for Sample B mixed with the colloidal silica. The gel strength of Sample B mixed with the colloidal silica is 25.9 kPa (3.75 pounds per square inch).

Claims

1. A composition capable of absorbing an aqueous liquid which composition comprises a functionally effective amount of a swellable addition polymer comprising in polymerized form,
 - (a) a major portion of at least one water-soluble, ionizable or hydrolyzable monomer, which is capable of forming at least a 10 weight percent solution in water and which readily undergoes addition polymerisation to form water-soluble polymers,
 - (b) a minor portion of at least one water-insoluble ethylenically unsaturated monomer having a pendant hydrophobic moiety where less than 0.4 weight parts of the

water-insoluble monomer will dissolve in 100 parts by weight water, and

(c) a minor portion of a crosslinking moiety in an amount sufficient to provide gel strength to the polymer with the proviso that the composition does not comprise an inter-polymer of an aliphatic glycol having 2 - 10 carbon atoms, and when the polymerized monomers comprise an acrylic ester with an alkyl group having 10 to 30 carbon atoms, the polymer does not comprise polymerized monomers of an acrylic acid ester with an alkyl group having 1 - 9 carbon atoms, or an acrylic or methacrylic nitrile or amide and which polymer is capable of undergoing hydrophobic association through hydrophobic groups of said polymer such that said polymer is swellable but not readily soluble in the presence of an aqueous liquid when said polymer is in intimate contact with an amount of surfactant, if any, which is less than that amount of surfactant which will solubilize said polymer in said aqueous liquid.

2. The composition of Claim 1 wherein said polymer comprises, in polymerized form, from 90 to 99.995 mole percent of at least one water-soluble monomer, and from 0.005 to 10 mole percent of at least one monomer having a pendant hydrophobic moiety.
3. The composition of Claim 1 wherein said polymer additionally comprises, in polymerized form, from 0 to 50,000 ppm of crosslinkable monomer.
4. The composition of Claim 3 wherein said crosslinkable monomer is a polyvinyl monomer.
5. A composition of Claim 1 wherein said monomer having a pendant hydrophobic moiety is derived from the reaction of alkanols having from 4 to 20 carbon atoms with an ethylenically unsaturated carboxylic acid.
6. A composition of Claim 1 wherein said monomer having a pendant hydrophobic moiety is an alkyl acrylate or alkyl methacrylate wherein alkyl has from 4 to 20 carbon atoms.
7. The composition of Claim 1 containing a functionally effective amount of an inorganic colloidal support.
8. The composition of Claim 7 comprising from 20 to 100 weight percent polymer and from 0 to 80 weight percent colloidal support, based

on the weight of the polymer and colloidal support.

Revisions

1. Composition capable d'absorber un liquide aqueux, laquelle composition comprend une quantité fonctionnellement efficace d'un polymère d'addition gonflable, comprenant, sous forme polymérisée,
 - (a) une partie prépondérante d'au moins un monomère hydrosoluble, ionisable ou hydrolysable, qui est capable de former au moins une solution à 10% en poids dans l'eau et qui subit facilement la polymérisation par addition pour former des polymères hydrosolubles,
 - (b) une partie mineure d'au moins un monomère insaturé éthylénique insoluble dans l'eau, ayant une fraction hydrophobe attachée, moins de 0,4 partie en poids du monomère insoluble dans l'eau se dissolvant dans 100 parties en poids d'eau, et
 - (c) une partie mineure d'une fraction réticulante, en une proportion suffisante pour fournir au polymère une consistance de gel, étant entendu que la composition ne contient pas un interpolymère d'un glycol aliphatique ayant 2-10 atomes de carbone, et que, lorsque les monomères polymérisés comprennent un ester acrylique comportant un groupe alkyle ayant 10 à 30 atomes de carbone, le polymère ne comprend pas des monomères polymérisés consistant en un ester d'acide acrylique avec un groupe alkyle ayant 1-9 atomes de carbone, ou en un nitrile ou amide acrylique ou méthacrylique, ledit polymère étant capable de subir une association hydrophobe par l'intermédiaire de groupes hydrophobes dudit polymère, de sorte que ledit polymère est gonflable mais n'est pas facilement soluble en présence d'un liquide aqueux lorsque ledit polymère est en contact intime avec une certaine quantité de tensioactif, s'il en existe, qui est inférieure à la quantité de tensioactif qui va solubiliser ledit polymère dans ledit liquide aqueux.
2. Composition selon la revendication 1, dans laquelle ledit polymère comprend, sous forme polymérisée, de 90 à 99,995 % en moles d'au moins un monomère hydrosoluble, et de 0,005 à 10 % en moles d'au moins un monomère ayant une fraction hydrophobe attachée.
3. Composition selon la revendication 1, dans laquelle ledit polymère comprend en outre, sous

forme polymérisée, de 0 à 50.000 ppm de monomère réticulable.

4. Composition selon la revendication 3, dans laquelle ledit monomère réticulable est un monomère polyvinylique.
5. Composition selon la revendication 1, dans laquelle ledit monomère ayant une fraction hydrophobe attachée résulte de la réaction d'alcanols ayant de 4 à 20 atomes de carbone avec un acide carboxylique insaturé éthylénique.
6. Composition selon la revendication 1, dans laquelle ledit monomère ayant une fraction hydrophobe attachée est un acrylate d'alkyle ou un méthacrylate d'alkyle dans lequel le groupe alkyle comporte de 4 à 20 atomes de carbone.
7. Composition selon la revendication 1, contenant une proportion fonctionnellement efficace d'un support colloïdal minéral.
8. Composition selon la revendication 7, comprenant de 20 à 100 % en poids de polymère et de 0 à 80 % en poids de support colloïdal, par rapport au poids du polymère et du support colloïdal.

Ansprüche

1. Zusammensetzung, die zur Absorption von wäßriger Flüssigkeit befähigt ist und eine funktionell wirksame Menge eines quellbaren Additionspolymer enthält, das in polymerisierter Form enthält
 - (a) einen Hauptanteil mindestens eines wasserlöslichen, ionisierbaren oder hydrolysierbaren Monomeren, das in der Lage ist, eine mindestens 10 gewichtsprozentige Lösung in Wasser auszubilden und das leicht ein wasserlösliches Polymer durch Additionspolymerisation bildet,
 - (b) einen geringen Anteil mindestens eines wasserunlöslichen, ethylenisch ungesättigten Monomeren mit anhängenden hydrophoben Gruppen, wobei sich weniger als 0,4 Gewichtsteile des wasserunlöslichen Monomeren in 100 Gewichtsteilen Wasser lösen, und
 - (c) einen kleinen Anteil einer vernetzenden Gruppe in einer ausreichenden Menge, um dem Polymer Gelfestigkeit zu verleihen, mit der Bedingung, daß die Zusammensetzung kein Mischpolymer eines aliphatischen Glycols mit 2 - 10 Kohlenstoffatomen enthält, und wenn die polymerisierten Mono-

meren einen Acrylester mit einer Alkylgruppe mit 10 - 30 Kohlenstoffatomen enthalten, das Polymer kein polymerisierten Monomeren eines Acrylsäureesters mit einer Alkylgruppe mit 1 - 9 Kohlenstoffatomen oder ein Acrylnitril oder Methacrylnitril oder -amid enthält,

und wobei das Polymer solche hydrophobe Verbindung über hydrophobe Gruppen des Polymeren eingehen kann, daß das Polymer quellbar, aber nicht leicht löslich, in Gegenwart einer wäßrigen Flüssigkeit ist, wenn das Polymer in innigem Kontakt mit einer Menge oberflächenaktiven Mittels steht, die in jedem Falle geringer ist als die Menge oberflächenaktiven Mittels, die das Polymer in der wäßrigen Flüssigkeit löslich macht.

2. Zusammensetzung nach Anspruch 1,

dadurch gekennzeichnet,

daß das Polymer in polymerisierter Form enthält von 90 - 99,995 Mol-% mindestens eines wasserlöslichen Monomeren und von 0.005 - 10 Mol-% mindestens eines Monomeren, das eine anhängende hydrophobe Gruppe aufweist.

3. Zusammensetzung nach Anspruch 1,

dadurch gekennzeichnet,

daß das Polymer in polymerisierter Form weiterhin enthält von 0 bis 50.000 ppm eines vernetzbaren Monomeren.

4. Zusammensetzung nach Anspruch 3,

dadurch gekennzeichnet,

daß das vernetzbare Monomer ein Polyvinylmonomer ist.

5. Zusammensetzung nach Anspruch 1,

dadurch gekennzeichnet,

daß das Monomer mit einer anhängenden hydrophoben Gruppe abgeleitet ist aus der Umsetzung eines Alkanols mit 4 - 20 Kohlenstoffatomen mit einer ethylenisch ungesättigten Carbonsäure.

6. Zusammensetzung nach Anspruch 1,

dadurch gekennzeichnet,

daß das Monomer mit einer anhängenden hydrophoben Gruppe ein Alkylacrylat oder Alkylmethacrylat ist, in dem Alkyl 4 - 20 Kohlenstoffatome aufweist.

7. Zusammensetzung nach Anspruch 1,

dadurch gekennzeichnet,

daß sie eine funktionell wirksame Menge eines anorganischen kolloidalen Trägers enthält.

8. Zusammensetzung nach Anspruch 7,

dadurch gekennzeichnet,

daß sie von 20 - 100 Gew.-% Polymer und von 0 - 80 Gew.-% kolloidalen Träger, bezogen auf das Gewicht des Polymeren und kolloidalen Trägers, enthält.